

the new metabolite. This observation, as well as chromatographical data, indicated that the new metabolite was acetylated methylhistamine or another hydrolysable conjugate of methylhistamine.

The radioactivity remaining in the fraction of the new metabolite after hydrolysis may be due to incomplete hydrolysis. However, the possibility also exists that at least part of the activity, but more than 2–3%, may stem from histaminol. By the methods used, quantitation of histaminol was impossible. To identify the unknown metabolite, it was necessary to separate it from N-acetylhistamine. The main fractions of the new metabolite were rechromatographed on an Aminex column until the metabolite seemed to move as a single compound.

The new metabolite is easily extractable from a basic solution with n-butanol and can be reextracted as histamine, from the organic solution with HCl. Butanol extraction followed by reextraction with HCl was combined with repeated column chromatography for isolation and purification. By starting with a 24-h urine, enough substance for direct mass-spectrographic analysis was obtained. The mass-spectra (figure 2a) indicate that the substance is N-acetyl-1.4-methylhistamine.

Reference N-acetylmethylhistamine was synthesized by methylation of N-acetylhistamine with dimethylsulfate and purified by means of TLC and butanol-HCl extractions. The purified compound could be visualized neither with ninhydrin nor by diazotation. The mass-spectrum for synthetic N-acetylmethylhistamine HCl (figure 2b) is nearly identical to that of the new metabolite isolated from the hens' urine. From this evidence, we conclude that N-acetyl-1.4-methylhistamine is a metabolite of histamine in chickens.

About 90% of the radioactivity could be accounted for by identified histamine-metabolites, the remaining 10%

may represent other metabolites such as fragments of the histamine-ring<sup>6</sup>. However, one cannot exclude the possibility that chromatographical effects may result in underestimation of some of the metabolites. The table demonstrates the metabolic pattern of exogenous histamine in chickens. The metabolic pattern of histamine in chicken differs largely from that found in other species. This difference is first and foremost a reflection of the new metabolite, N-acetylmethylhistamine, which accounted for about 20% of the excreted radioactivity, but also of the great quantitative importance of N-acetylhistamine. The identity of N-acetylhistamine was confirmed with mass-spectrography.

The present experiments support the conclusion of Shifrine et al.<sup>3</sup> that acetylation is of great importance in the detoxication of histamine in chicken. This might apply to fowls generally since in vitro experiments with liver slices and cell free extracts of pigeon liver are potent to conjugate histamine<sup>7</sup>.

The fact that Bergmark and Granerus<sup>4</sup> found only very low radioactivity in the fractions where N-acetylmethylhistamine should occur suggests that this metabolite, quantitatively important in chickens, is negligible, if present at all, in man.

Whether acetylation or methylation of histamine is the first step in the formation of N-acetylmethylhistamine, and in which tissues these reactions take place, is under examination. Since neither N-acetylhistamine nor 1.4-methylhistamine possess biological activity, it seems likely that this is true also for N-acetyl-1.4-methylhistamine.

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## Reduction of adenylylsulfate and 3'-phosphoadenylylsulfate in phototrophic bacteria

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**Summary.** Extracts of 14 species of phototrophic bacteria, partly grown with different sulfur compounds, were tested for their ability to form volatile sulfur compounds from adenylylsulfate (APS) and 3'-phosphoadenylylsulfate (PAPS). The *Rhodospirillum* species showed marked activities with both APS and PAPS while the *Rhodopseudomonas* species seem to prefer PAPS. The Chromatiaceae exhibited the strongest activities with APS, whereas *Chlorobium limicola* had equally high activity with PAPS.

During dissimilatory sulfur metabolism in Chlorobiaceae and Chromatiaceae, sulfite is oxidized by APS<sup>2</sup> reductase<sup>3</sup>. The relatively high levels of APS reductase found in *Chromatium vinosum* and *Thiocapsa roseopersicina* after photoheterotrophic growth with sulfate, as the sole sulfur source, led to the assumption that APS reductase might be involved also in assimilatory sulfur metabolism in these bacteria<sup>4</sup>, i.e. in the reduction of APS to sulfite. On the other hand, none of the *Rhodospirillaceae* contain APS reductase<sup>4</sup>, and the formation of PAPS<sup>2</sup> from sulfate and ATP has been reported to occur in chromatophores of *Rhodospirillum rubrum*<sup>5</sup>.

The aim of this study was to test a number of representative species of the phototrophic bacteria as to whether they are able to form volatile sulfur compounds from PAPS or APS (or both).

**Material and methods.** The *Rhodospirillaceae* were grown on conventional media<sup>6,7</sup> with sulfate, or special media<sup>8,9</sup> with reduced sulfur compounds. The *Chlorobium* strain and the Chromatiaceae were cultivated in Pfennig's

medium<sup>10</sup> with sulfide. *C. vinosum* also photoheterotrophically. Incubation occurred in 500 ml screw-capped bottles at 25–30°C and about 2000 lux.

Frozen cells were thawed and suspended in a buffer containing 0.1 M Tris-HCl, pH 8.0, 10 mM MgCl<sub>2</sub>, 0.1 M KCl,

1 We thank J. F. Imhoff for skilful assistance and the Deutsche Forschungsgemeinschaft for financial support.

2 Abbreviations. APS, Adenylylsulfate (= adenosine phosphosulfate). DTE, Dithioerythritol. PAPS, 3'-Phosphoadenylylsulfate.

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and 5 mM DTE<sup>2</sup>. 2 ml of buffer were used for 1.0 g of wet weight cells. These were broken at 10,000 psi (French press) and centrifuged to remove unbroken material. Protein was determined by the biuret method after trichloroacetic acid precipitation and 2fold acetone extraction to remove pigments<sup>11</sup>. Preparation of APS and PAPS was performed using the slightly modified *Chlorella* system<sup>12, 13</sup>.

Table 1. Reduction of sulfonucleotides in crude (chromatophore-containing) extracts of phototrophic bacteria

Species	Strain	Sulfur source	nmols sulfonucleotide reduced per mg protein/h	
			APS	PAPS
<i>Rhodospirillum rubrum</i>	1761-1a	Sulfate	0.234	0.453
<i>fulvum</i>	BN211	Sulfate	0.168	0.173
(chromatophore fraction alone)			0.020	0.066
(supernatant alone)			0.126	0.554
<i>tenuis</i>	BN230	Sulfate	0.206	0.493
		Cysteine	0.372	0.468
<i>photometricum</i>	UW	Sulfate	0	0
<i>Rhodopseudomonas palustris</i>	DSM123*	Thiosulfate	0	0.017
	DSM127	Sulfide	0	0.113
		Thiosulfate	0	0.261
	BN128*	Sulfate	0.012	0.165
<i>viridis</i>	BN170	Sulfate	0.006	0.461
<i>capsulata</i>	DSM155	Sulfate	0.099	0.145
		Sulfide	0.006	0.019
<i>gelatinosa</i>	DSM149	Sulfate	0.019	0.029
<i>globiformis</i>	DSM161	Cysteine	0	0
		Thiosulfate	0.003	0.004
		Sulfate	0.005	0.009
<i>sulfidophila</i>	W4	Sulfate	0	0.023
		Sulfide	0	0.001
	BN982*	Sulfide	0	0
<i>Thiocapsa roseopersicina</i>	DSM219	Sulfide	4.720	0.447
<i>pfennigii</i>	DSM226	Sulfide	0.700	0.012
<i>Chromatium vinosum</i>	DSM180	Sulfate	0.174	0.074
		Sulfide	0.571	0.287
<i>Chlorobium limicola</i>	DSM249	Sulfide	0.640	0.745

Conditions see table 2.

\*KCl omitted from buffer mixture during preparation of extract.

Table 2. Isotope exchange reaction between labelled sulfonucleotides and unlabelled sulfite catalyzed by bacterial extracts

Organism	Formation of radioactive acid volatile sulfur (nmols/h × mg protein) from AP <sup>35</sup> S	PAP <sup>35</sup> S
<i>R. rubrum</i>	0.005	0.027
<i>Chr. vinosum</i>	0.32	0.013
<i>Tca. pfennigii</i>	1.16	0.14
<i>Chl. limicola</i>	1.58	1.04

Conditions (in  $\mu$ moles): Tris-HCl, pH 9.0: 100; MgCl<sub>2</sub>: 10; unlabelled sulfite: 100; AP<sup>35</sup>S: 0.035 (1 nmole  $\triangleq$  1891 cpm) when indicated; PAP<sup>35</sup>S: 0.035 (1 nmole  $\triangleq$  1315 cpm) when indicated; bacterial extracts, total volume 1 ml. After 1 h at 37°C under N<sub>2</sub>, 0.1 ml of 1 M SO<sub>3</sub><sup>2-</sup> was added and the acid volatile radioactivity trapped in 1 ml 1 M triethanolamine. The radioactivity was determined by scintillation counting<sup>14</sup> in a Packard Tricarb.

In order to obtain useful concentrations for a reaction mixture suitable to study volatile sulfur formation from labelled sulfonucleotides, a pilot study with *R. rubrum* was made. The optimal pH for PAPS reduction was 9.0. The relationship between reaction rate and protein concentration was linear up to about 25 mg of protein per test volume. From the influence of the PAPS concentrations, an apparent K<sub>m</sub>-value of  $4 \cdot 10^{-5}$  M for PAPS was obtained by Lineweaver-Burk plot.

A suitable reaction mixture was prepared from these data, buffered at pH 9.0, containing 35  $\mu$ moles of labelled sulfonucleotide and 1–20 mg of protein per ml.

**Results.** Table 1 shows the result of 54 incubations carried out with extracts of 14 different species of phototrophic bacteria partly grown on different sulfur sources. The highest activities with APS were found in those bacteria that are known to contain APS reductase, the Chromatiaceae and *Chlorobium*. But these bacteria also showed relatively high activity with PAPS. Another group exhibiting fairly high activity with APS is found in the *Rhodospirillum* species (except *R. photometricum*), where reactivity with PAPS was only slightly higher. The lowest activities – often down to zero – for APS occur in the *Rhodopseudomonas* species tested, where the activity with PAPS was always higher, especially when the cells had been grown with sulfate.

The experiments were carried out with chromatophore-containing crude extracts that were only freed from unbroken cells, large fragments and sulfur globules by low speed centrifugation. It is, however, most unlikely that the extracts still contained or produced (by photophosphorylation) sufficient ATP to allow a phosphorylation of APS to PAPS by APS kinase. This is also supported by the comparison made with chromatophore-free fractions (table 1) and controls in the dark (data not shown). On the contrary, the apparent PAPS reduction also occurring in the Chromatiaceae and in *Chlorobium* is most likely due to the presence of a 3'-nucleotidase in the extracts, leading to an intermediary APS formation from PAPS. This assumption is supported by the isotope exchange reaction between labelled APS (or PAPS) and unlabelled sulfite (table 2). If a regular APS reductase is present, together with a small amount of AMP, one would expect reactivity in both directions<sup>15, 16</sup>: In reaction mixtures containing <sup>35</sup>S-sulfonucleotide and unlabelled sulfite, the latter together with AMP would form unlabelled APS. This would mix with labelled APS and react back forming labelled sulfite. Thus, label in sulfite (i.e., volatile sulfur) will increase. Again, a 3'-nucleotidase would facilitate PAPS to participate via APS. This has to be postulated, because APS reductases have been shown to be specific for APS<sup>16</sup>. Table 2 shows the results of such isotope exchange experiments.

The organisms known to contain APS reductase show reasonable exchange rates, while in *R. rubrum* no substantial exchange takes place. This proves that a different, non-reversible mechanism, such as occurs in all assimilatory sulfate reducers so far studied, is present also in this species.

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The results with *R. rubrum*, and probably the other *Rhodospirillum* species (except *R. photometricum*), point to the possibility that in these bacteria an APS sulfotransferase like that found in algae and higher plants<sup>13,17,18</sup> is active. We know that our results do not present final proof for this possibility, and also that we

cannot exclude the presence of APS sulfotransferase besides APS reductase in the phototrophic sulfur bacteria. Further studies with purified enzymes are under way.

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## Effect of trypsin, S-adenosylmethionine and ethionine on L-serine sulphydrase activity<sup>1</sup>

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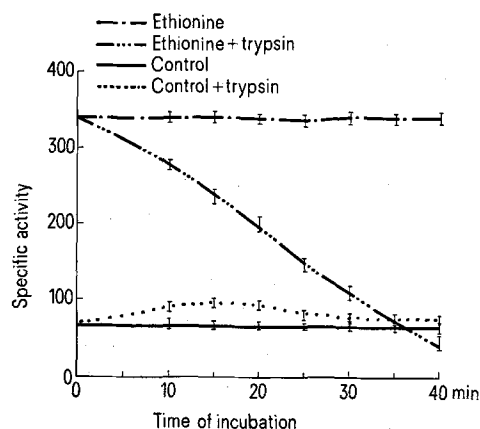
**Summary.** Trypsin causes an activation of serine sulphydrase in the liver extracts from intact animals, but inhibits enzyme activity in the liver of ethionine treated rats. Trypsin also decreases an elevation of serine sulphydrase activity caused by S-adenosylmethionine.

During the last few years, several studies have indicated that both serine sulphydrase (EC 4.2.1.22) and cystathionine  $\beta$ -synthetase activities are due to a single enzyme<sup>2-4</sup>. Recently, it has been demonstrated that ethionine administration produces a significant activation of cystathionine  $\beta$ -synthetase and serine sulphydrase in the rat tissues<sup>4-6</sup>. It is also found that S-adenosylmethionine affects as the activator of cystathionine  $\beta$ -synthetase<sup>4,6</sup>. Mudd and his coworkers<sup>7</sup> have shown that trypsin increases the activity of cystathionine-synthetase from an extract of human liver.

As the control mechanism of serine sulphydrase action has not been known of sufficiently, we examined the interaction of trypsin, ethionine and S-adenosylmethionine in relation to the activity of this enzyme.

Effects of S-adenosylmethionine (SAM) and trypsin on L-serine sulphydrase activity in vitro (nmoles cysteine/mg protein h)

No. of experiments	Medium for incubation (20 min at 37°C)	Enzyme activity
9	Enzyme + SAM	99.6 $\pm$ 1.8
9	Enzyme + SAM + trypsin	81.5 $\pm$ 1.4
9	Enzyme + trypsin	97.3 $\pm$ 2.1
9	Enzyme	76.1 $\pm$ 0.9



In vitro effect of trypsin on L-serine sulphydrase activity in the liver of rats treated with ethionine. Vertical bars indicate standard errors of the mean.

**Material and methods.** Albino rats weighing 160–220 g were used. DL-ethionine was dissolved in 0.9% NaCl and injected i.p. in a dose of 200 mg/kg. The control group received the corresponding volume of 0.9% NaCl. The animals were killed 3 h after the injection. Serine sulphydrase was extracted and separated from serine dehydratase according to the Kashiwamata and Greenberg method<sup>8</sup>. Trypsin was dissolved in 0.1 M tris-HCl buffer pH 8.3 and added to the aliquots of enzyme extract in the amount of 0.5 y/mg protein. S-adenosylmethionine was dissolved in H<sub>2</sub>O and added to the enzyme of control rat liver in the amount of 0.1 micromol per mg protein. Enzyme had been preincubated with S-adenosylmethionine 5 min at 37°C before adding trypsin. After incubation at 37°C, the activity of serine sulphydrase was determined by the procedure of Stepien and Pieniazek<sup>9</sup>. Enzyme activity was expressed in nmoles cysteine per mg protein per h. Protein was estimated by the method of Lowry et al.<sup>10</sup>. DL-ethionine and S-adenosylmethionine were obtained from Calbiochem while trypsin was purchased from Merck.

**Results and discussion.** The effect of trypsin treatment on L-serine sulphydrase activity is shown in the figure. Trypsin causes an increase in serine sulphydrase activity in the liver extracts from intact animals. On the contrary, when trypsin was incubated with enzyme extracts prepared from the liver of the rats which had received DL-ethionine, a successive diminution in serine sulphydrase activity was observed. The action of S-adenosylmethionine and trypsin upon serine sulphydrase levels is presented in the table. It is evident from these data that S-adenosylmethionine as well as trypsin produce a remarkable rise in serine sulphydrase. However, addition

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